

ZITHOLELE CONSULTING (PTY) LTD

**WASTE CLASSIFICATION OF POWER STATION
ASH AND BRINE FROM THE CAMDEN POWER STATION**

Report No.: JW164/11/D116 - REV 6

September 2014



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Acronyms and abbreviations used in this document:

ARL	Acceptable Risk Level. (ARL = 0.1 x LC ₅₀)
ARLP	South African Acid Rain Leach Procedure
ASLP	Australian Standard Leaching Procedure
DEA	Department of Environmental Affairs
DWA	Department of Water Affairs
DWAF	Department of Water Affairs and Forestry
G:L:B⁺	General waste landfill receiving more than 500 tonnes of waste per day with a barrier system containing a leachate detection and collection layer
H:H	Hazardous waste disposal facility suitable for the disposal of all Hazard Group 1, 2, 3, 4 and general wastes. Comply with the most conservative design as indicated in the DWAF's Minimum Requirements
H:h	Hazardous waste disposal facility suitable for the disposal of all Hazard Group 3 and 4 wastes, and general wastes. Comply with the second most conservative design as indicated in the DWAF's Minimum Requirements
LC	Leach concentration in mg/ℓ
LCT	Leach concentration threshold in mg/ℓ
LC₅₀	The concentration at which 50% of test organisms will die after a certain exposure time
mg/kg	Milligram per kilogram
mg/ℓ	Milligram per litre
RO	Reverse osmosis
TC	Total concentration in mg/kg
TCT	Total concentration threshold
TCLP	Toxic characteristic leach procedure
TDS	Total dissolved salts
μS/cm	Micro Siemens per centimetre

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SGS SOUTH AFRICA: LABORATORY CERTIFICATES

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CHEMICAL ANALYSES CONDUCTED ON THE REVERSE OSMOSIS BRINE



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1. INTRODUCTION

1.1 Background

Zitholele Consulting (Pty) Ltd is currently in the process of conducting an Environmental Impact Assessment (EIA) and Waste Licence Application for a new wet ash disposal facility at the Camden Power Station. The new ash disposal site will be approximately 100 hectares in size with a further 25 hectares for associated infrastructure. The power station also operates a Reverse Osmosis (RO) plant in order to reduce the positive water balance. This plant generates a brine and the brine is currently co-disposed with the wet ash on the existing ash disposal facility.

The classification of the ash from the wet-ash deposition process at Camden Power Station is required for input into both the EIA and Waste Licence Application Report. In addition, the ash classification is required to determine its environmental risk profile and hence the barrier design criteria applicable to the new ash disposal facility. Classification of the brine is also required in order to establish its risk profile.

The ash was originally classified in terms of both the Department of Water Affairs and Forestry's (DWA's) "Minimum Requirements for the Handling, Classification and Disposal of Hazardous Waste" of 1998 (DWA, 1998a) and the Department of Environmental Affairs' draft "National Environmental Management: Waste Act (Act 59 of 2008). Draft Standard for Assessment of Waste for Landfill Disposal" (DEA, 2011). The outcome of this classification is dealt with in Jones & Wagener's report no JW164/11/D116 - REV 3 dated September 2012.

In January 2014 J&W was requested by Zitholele Consulting to update the classifications based on the DEA's "National Norms and Standards for the Assessment of Waste for Landfill Disposal" (National Norms and Standards) (DEA, 2013a). The National Norms and Standards were promulgated in August 2013 and replaced the Minimum Requirements waste classification system.

1.2 Objectives

The objective was to reclassify the Camden Power Station's wet ash and RO plant brine in terms of the DEA's Norms and Standards of 2013. The analytical results of the tests performed in 2012 on the wet ash were used for this classification. The original classification of the brine was based on theoretical values provided, but for this revised classification, chemical analyses were made available for some of the constituents listed in the National Norms and Standards.

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2. DEA WASTE CLASSIFICATION SYSTEM

The new waste classification system, which replaced the Department of Water Affairs' Minimum Requirements classification system on 23 August 2013, focuses on the long term storage (in excess of 90 days) and disposal of waste on land or in waste disposal facilities. The system is based on the Australian State of Victoria's waste classification system for disposal, which uses the Australian Standard Leaching Procedure (ASLP) to determine the leachable concentrations (LCs) of pollutants (DEA, 2013a).

For waste to be disposed of with putrescible organic matter, an acetic acid leach solution is used. This leach solution is very similar to the US EPA TCLP leach solution used in the now outdated Minimum Requirements, except that the pH is 5.0, instead of pH 4.93. In cases where a waste has a high pH, and following an acid neutralisation capacity test, a pH 2.9 leach solution must be used.

In cases where non-organic waste, such as the power station ash, is to be co-disposed with other non-organic waste, a basic 0.10 M sodium tetraborate decahydrate (borax) solution of pH 9.2 ± 0.10 should be used in addition to the acetic acid leach (DEA, 2012a). The objective of the sodium tetraborate test is to identify contaminants that are leached above the various leachable concentration thresholds (LCTs) trigger values at a high pH¹.

For non-putrescible inorganic waste to be disposed of without any other wastes (mono-disposal scenario), reagent water (distilled water) is used as a leach agent.

In addition to the above, the TCs of the constituents of concern need to be determined and compared to specified total concentration threshold (TCT) values (DEA, 2013a)².

The number of potentially hazardous substances in the new classification system has been significantly reduced from that listed in the old Minimum Requirements of 1998 and brought in line with the potentially hazardous substances being used in other parts of the world to classify waste for disposal purposes. However, if a generator is aware of a hazardous substance other than those listed by the DEA, they are obliged to indicate and analyse for this.

Once the analytical results are known, the waste is classified in line with the following approach:

- Wastes with any element or chemical substance concentration above the LCT3 or TCT2 values ($LC > LCT3$ or $TC > TCT2$) are Type 0 Wastes. Type 0 wastes (extremely hazardous waste), require treatment/stabilisation before disposal;
- Wastes with any element or chemical substance concentration above the LCT2 but below LCT3 values, or above the TCT1 but below TCT2 values ($LCT2 < LC \leq LCT3$ or $TCT1 < TC \leq TCT2$), are Type 1 Wastes (highly hazardous waste, which must be

¹ LCT1 limits have, where possible, been derived from the lowest value of the standard for human health effects listed for drinking water (LCTO) in South Africa (DWAf, SANS) by multiplying with a Dilution Attenuation Factor (DAF) of 50 as proposed by the Australian State of Victoria, "Industrial Waste Resource Guidelines: Solid Industrial Waste Hazard Categorisation and Management", June 2009 (www.epa.vic.gov.au). If no standard was available in South Africa then the limits given by the WHO or other appropriate drinking water standard, such as those published in the California Regulations have been used.

LCT2 limits were derived by multiplying the LCT1 value with a factor of 2, and the LCT3 limits have been derived by multiplying the LCT2 value with a factor of 4. The factors applied represents a conservative assessment of the decrease in risk achieved by the increase in environmental protection provided by more comprehensive liner designs in higher classes of landfill and landfill operating requirements.

² TCT1 limits were derived from the land remediation values for commercial/industrial land determined by the Department of Environmental Affairs' "Framework for the Management of Contaminated Land". The TCT2 limits were derived by multiplying TCT1 by a factor of 4, as used by the Environmental Protection Agency, Australian State of Victoria



disposed of on a Class A landfill constructed with the most conservative barrier system);

- Wastes with any element or chemical substance concentration above the LCT1 but below the LCT2 values and all concentrations below the TCT1 values ($LCT1 < LC \leq LCT2$ and $TC \leq TCT1$) are Type 2 Wastes (moderate hazardous waste, which must be disposed of on a Class B landfill);
- Wastes with any element or chemical substance concentration above the LCT0 but below LCT1 values and all concentrations below the TCT1 values ($LCT0 < LC \leq LCT1$ and $TC \leq TCT1$) are Type 3 Wastes (low hazardous waste, which must be disposed of on a Class C landfill);
- Wastes with all elements and chemical substance concentration levels for metal ions and inorganic anions below the LCT0 and TCT0 values ($LC \leq LCT0$ and $TC \leq TCT0$), as well as below the limits for organics and pesticides as in **Table 2-1**, are Type 4 Wastes (near inert wastes, which must be disposed of on sites with some base preparation, but no formal barrier system):

Table 2-1: Organic limits for wastes to be classified as Type 4 wastes.

Chemical Substances in Waste	Total Concentration (mg/kg)
Organic constituents	
Total organic carbon (TOC)	30 000 (3%)
Benzene, toluene, ethyl benzene and xylenes (BTEX)	6
Polychlorinated Biphenyls (PCBs)	1
Mineral Oil (C10 to C40)	500
Pesticides	
Aldrin + Dieldrin	0.05
DDT + DDD + DDE	0.05
2,4-D	0.05
Chlordane	0.05
Heptachlor	0.05

- Wastes with all element or chemical substance leachable concentration levels for metal ions and inorganic anions below or equal to the LCT0 limits are considered to be Type 3 waste, irrespective of the total concentration of elements or chemical substances in the waste, provided that:
 - All chemical substance concentration levels are below the total concentration limits for organics and pesticides in the **Table 2-1**;
 - The inherent physical and chemical character of the waste is stable and will not change over time; and,
 - The waste is disposed of to landfill without any other waste.



- Wastes with the TC of an element or chemical substance above the TCT2 limit, and where the concentration cannot be reduced to below the TCT2 limit, but the LC for the particular element or chemical substance is below the LCT3 limit, the waste is considered to be Type 1 Waste.

3. **TESTS CONDUCTED**

Camden Power Station supplied representative samples of dry ash, wet ash (2 samples) and ash disposal site leachate (seepage water) – see **Photo 1**. The samples were then sent to the SGS Laboratory in Randburg for various leach analyses, total concentration (TC) determination and quantitative x-ray diffraction (XRD) analysis to determine the mineralogy.

The SGS laboratory subjected the dry ash to a Minimum Requirements' Acid Rain Leach Procedure (ARLP). The ARLP leach procedure was used in the 1998 Minimum Requirements waste classification system where a waste is mono-disposed or stored or where it is co-disposed with other inorganic waste types not containing any decomposable compounds.

The dry ash sample was also subjected to a total extraction procedure in order to determine the TCs of the various elements.

In addition, the dry ash sample was subjected to a XRD analysis to determine the mineralogy.

Following the new DEA classification system for the mono storage and disposal of a waste, solids were firstly separated from the liquid fraction and the percentage solids determined. The solids fractions were then subjected to a deionised (DI) (South African Standard Leach Procedure) water leach test, where after the leach solution was analysed for various metals and other inorganic constituents. The water fractions of the two wet ash samples were also analysed for the various metals and inorganic constituents listed in the National Norms and Standards. The organic components listed in the National Norms and Standards were not analysed for as it is highly unlikely that organics will occur in the wet ash at concentrations above the LCT0 and TCT0 values of the National Norms and Standards.

The two wet ash samples provided were termed dusting ash, that is the fine ash-water mixture used to develop the outer walls of the current ash disposal facility and ashing ash, the coarse ash-water mixture. The coarse ash is deposited in the middle of the ash disposal facility. It is noted that the brine from the reverse osmosis plant is co-disposed with the wet ash.

A sample of leachate collected at the toe of the ash disposal facility (seepage water) was also analysed for various inorganic constituents.

The certificates of the results of the various tests conducted on the ash and leachate are included in **Appendix A**.

Although a sample of brine from the reverse osmosis (RO) plant was requested for analyses at the time, the plant was not operative on the day that the wet ash samples were collected. Theoretical values for the various constituents of concern were provided by Eskom Camden Power Station and these values were used in the initial classification. However, for this classification, Mrs I. Hodgson of the Camden Power Station provided some analyses performed on the RO plant brine to J&W on 20 February 2014 and also determined the conductivity of the brine on 20 February 2014 – see **Appendix B**. The conductivity of the brine was verbally reported as 3 309 $\mu\text{S}/\text{cm}$ (330.9 mS/m). For the classification of the brine, the 70% water recovery rate results were used, which provides



a more concentrated brine, therefore the more conservative scenario was used for the classification.

For the classification of the wet ash in terms of the DEA's National Norms and Standards the analytical results from the ARLP were ignored. Only the results obtained from the DI water leach and the TCs were used for the classification of the wet ash.



Photo 1: Four samples used in the classification of the Camden Power Station Ash, Ash Carrier Water and Ash Disposal Facility Seepage Water (Leachate)

4. CAMDEN POWER STATION ASH AND REVERSE OSMOSIS BRINE CLASSIFICATION

4.1 Wet Ash Classification

In order to determine the classification of the wet dusting ash (fine ash) and wet ashing ash (coarse ash) (both containing brine from the reverse osmosis plant), the percentage contributions of the concentrations of the constituents in the liquid fractions and the leach concentrations were calculated based on the percentage liquids to solids – see **Table 4-1** and **Table 4-3**. The corrected concentrations were then used for the classification – see **Table 4-2** and **Table 4-4**. Based on the corrected concentrations, both the dusting and ashing ash is classified as Type 3 wastes.

In addition, the concentrations of the listed constituents were also determined on the ash seepage water collected at the base of the existing ash disposal facility. Based on these concentrations, the ash is also classified as a Type 3 waste. It is noted that the TDS of the seepage water (764 mg/l) is lower than the average TDS of the dusting and ashing ash (1 424 mg/l).

Type 3 wastes should be disposed of on waste disposal facilities with a Class C landfill barrier system.

4.2 RO Plant Brine Classification

Based on the theoretical and actual concentrations provided for the RO plant brine, the brine is classified as a Type 3 liquid waste – see **Table 4-6**. The brine is classified as a Type 3 waste due to the concentrations of TDS, chloride, sulphate, fluoride, lead, total chromium and molybdenum being above their respective LCT0 values. Note that actual values were available for some of the constituents, but those marked red in **Table 4-6** are theoretical values supplied by Eskom.



Table 4-1: Corrected concentrations for dusting ash sample based on % contribution of ash carrier water and fine ash (dusting ash) content

DUSTING SAMPLE							
Percentage solids	48.30%						
WATER LEACH: DUSTING SAMPLE							
	Solid Phase			Water Phase			Leach Concentration
Element/Compound	mg/ℓ	Contribution Factor	Corrected concentration in mg/ℓ	mg/ℓ	Contribution Factor	Corrected concentration in mg/ℓ	mg/ℓ
As, Arsenic	0.0015	0.483	0.0007245	0.0015	0.517	0.0007755	0.0015
B, Boron	0.2	0.483	0.0966	0.11	0.517	0.05687	0.15347
Ba, Barium	0.84	0.483	0.40572	1.3	0.517	0.6721	1.07782
Cd, Cadmium	0.001	0.483	0.000483	0.001	0.517	0.000517	0.001
Co, Cobalt	0.001	0.483	0.000483	0.001	0.517	0.000517	0.001
Cr, Chromium - total	0.11	0.483	0.05313	0.15	0.517	0.07755	0.13068
Cr VI, Chromium VI	0.11	0.483	0.05313	0.15	0.517	0.07755	0.13068
Cu, Copper	0.002	0.483	0.000966	0.002	0.517	0.001034	0.002
Hg, Mercury	0.0003	0.483	0.0001449	0.00005	0.517	0.00002585	0.00017075
Mn, Manganese	0.0015	0.483	0.0007245	0.0015	0.517	0.0007755	0.0015
Mo, Molydenum	0.067	0.483	0.032361	0.19	0.517	0.09823	0.130591
Ni, Nickel	0.0035	0.483	0.0016905	0.0035	0.517	0.0018095	0.0035
Pb, Lead	0.002	0.483	0.000966	0.002	0.517	0.001034	0.002
Sb, Antimony	0.0035	0.483	0.0016905		0.517	0	0.0016905
Se, Selenium	0.002	0.483	0.000966	0.002	0.517	0.001034	0.002
V, Vanadium	0.045	0.483	0.021735	0.0021	0.517	0.0010857	0.0228207
Zn, Zinc	0.005	0.483	0.002415	0.005	0.517	0.002585	0.005
TDS, Total dissolved salts	272	0.483	131.376	1992	0.517	1029.864	1161.24
Cl, Chloride	2.1	0.483	1.0143	120	0.517	62.04	63.0543
SO ₄ , Sulphate	13	0.483	6.279	210	0.517	108.57	114.849
NO ₃ , Nitrate	1.5	0.483	0.7245	0.64	0.517	0.33088	1.05538
F, Fluoride	0.3	0.483	0.1449	0.73	0.517	0.37741	0.52231

Note: In order to calculate the % contibution of each phase, values less than (<) the limit of report (LOR) were divided by 2

Table 4-2: De-ionised Water Leach Test Results of Camden Power Station Ash (TC Dry Ash, LC Dusting sample)

Camden Power Station Ash: Dusting Ash																	
Chemical Species	Deionised Water Leach (LC) mg/ℓ	Total Concentration (TC) mg/kg	Limit of Report for LC mg/ℓ	LCT0		LCT1		LCT2		LCT3		TCT0		TCT1		TCT2	
				mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg
				TYPE 4 WASTE		TYPE 3 WASTE		TYPE 2 WASTE		TYPE 1 WASTE		TYPE 0 WASTE		TYPE 0 WASTE		TYPE 0 WASTE	
As	0.0015	13	0.0030	0.010	5.8	0.50	500	1.0	500	4.0	2 000						
B	0.15	NA	0.220	0.50	150	25	15 000	50	15 000	200	60 000						
Ba	1.1	716	0.030	0.70	62.5	35	6 250	70	6 250	280	25 000						
Cd	0.0010	<0.020	0.0020	0.003	7.5	0.15	260	0.30	260	1.2	1 040						
Co	0.0010	16	0.0020	0.50	50	25	5 000	50	5 000	200	20 000						
Cr	0.13	113	0.040	0.10	46 000	5.0	800 000	10	800 000	40							
Cr(VI)	0.13	NA	0.010	0.050	6.5	2.5	500	5.0	500	20	2 000						
Cu	0.0020	59	0.0040	2.0	16	100	19 500	200	19 500	800	78 000						
Hg	0.00017	<3.0	0.00010	0.006	0.93	0.30	160	0.6	160	2.4	640						
Mn	0.0015	488	0.060	0.50	1 000	25	25 000	50	25 000	200	100 000						
Mo	0.13	5.2	0.020	0.070	40	3.5	1 000	7.0	1 000	28	4 000						
Ni	0.0035	51	0.0070	0.070	91	3.5	10 600	7.0	10 600	28	42 400						
Pb	0.0020	41	0.0040	0.010	20	0.50	1 900	1.0	1 900	4.0	7 600						
Sb	0.0017	0.89	0.0070	0.02	10	1.00	75	2.00	75	8.00	300						
Se	0.0020	<2.0	0.0040	0.010	10	0.50	50	1.0	50	4.0	200						
V	0.023	68	0.0030	0.20	150	10	2 680	20	2 680	80	10 720						
Zn	0.0050	314	0.080	5.0	240	250	160 000	500	160 000	2 000	640 000						
TDS	1 161		21	1 000		12 500		25 -000	N/A	100 000	N/A						
Chloride	63		0.50	300		15 000		30 000	N/A	120 000	N/A						
Sulphate as SO ₄	115		0.40	250		12 500		25 000	N/A	100 000	N/A						
NO ₃ as N	1.1		0.40	11		550		1 100	N/A	4 400	N/A						
Fluoride	0.52	NA	0.30	1.5	100	75	10 000	150	10 000	600	40 000						
NA	Not analysed																
N/A	Not available																
	LC ≤ LCT0 and TC ≤ TCT0: Type 4 wastes																
	LCT0 < LV ≤ LCT1 and TC ≤ TCT1: Type 3 Wastes																
	LCT1 < LC ≤ LCT2 and TC ≤ TCT1: Type 2 Waste																
	LCT2 < LC ≤ LCT3 or TCT1 < TC ≤ TCT2: Type 1 Wastes																
	LC > LCT3 or TC > TCT2: Type 0																



Table 4-3: Corrected concentrations for ashing sample based on % contribution of ash carrier water and ashing (coarse) ash content

ASHING SAMPLE (Wet)							
Percentage solids	6.37%						
WATER LEACH: ASHING SAMPLE							
Element/Compound	Solid Phase			Water Phase			Leach Concentration
	mg/ℓ	Contribution Factor	Corrected concentration in mg/ℓ	mg/ℓ	Contribution Factor	Corrected concentration in mg/ℓ	mg/ℓ
As, Arsenic	0.012	0.064	0.00076	0.0015	0.9363	0.0014	0.0022
B, Boron	0.39	0.064	0.025	1.1	0.9363	1.03	1.1
Ba, Barium	0.059	0.064	0.0038	0.34	0.9363	0.32	0.32
Cd, Cadmium	0.0024	0.064	0.00015	0.0010	0.9363	0.00094	0.0011
Co, Cobalt	0.0027	0.064	0.00017	0.0010	0.9363	0.00094	0.0011
Cr, Chromium - total	0.0075	0.064	0.00048	0.029	0.9363	0.027	0.028
Cr VI, Chromium VI	0.0050	0.064	0.00032	0.030	0.9363	0.028	0.028
Cu, Copper	0.0020	0.064	0.00013	0.0020	0.9363	0.0019	0.0020
Hg, Mercury	0.00015	0.064	0.0000096	0.0012	0.9363	0.0011	0.0011
Mn, Manganese	0.0097	0.064	0.00062	0.0015	0.9363	0.0014	0.0020
Mo, Molybdenum	0.012	0.064	0.00076	0.18	0.9363	0.17	0.17
Ni, Nickel	0.0035	0.064	0.00022	0.0035	0.9363	0.0033	0.0035
Pb, Lead	0.0020	0.064	0.00013	0.0020	0.9363	0.0019	0.0020
Sb, Antimony	0.0035	0.064	0.00022		0.9363	0	0.00022
Se, Selenium	0.0020	0.064	0.00013	0.0094	0.9363	0.0088	0.0089
V, Vanadium	0.022	0.064	0.0014	0.020	0.9363	0.019	0.020
Zn, Zinc	0.0050	0.064	0.00032	0.0050	0.9363	0.0047	0.0050
TDS, Total dissolved solids	64	0.064	4.1	856	0.9363	801	806
Cl, Chloride	1.7	0.064	0.11	97	0.9363	91	91
SO ₄ , Sulphate	19	0.064	1.2	380	0.9363	356	357
NO ₃ , Nitrate	0.28	0.064	0.018	3.2	0.9363	3.0	3.0
F, Fluoride	0.025	0.064	0.0016	0.74	0.9363	0.69	0.69

Note: In order to calculate the % contribution of each phase, values less than (<) the limit of report (LOR) were divided by 2

Table 4-4: De-ionised Water Leach Test Results of Camden Power Station Ash (TC Dry Ash, LC Ashing sample)

Camden Power Station Ash: Ashing Sample															
Chemical Species	Deionised Water Leach (LC) mg/ℓ	Total Concentration (TC) mg/kg	Limit of Report for LC mg/ℓ	LCT0		TCT0		LCT1		TCT1		LCT2		TCT2	
				mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg		
				TYPE 4 WASTE		TYPE 3 WASTE		TYPE 2 WASTE		TYPE 1 WASTE		TYPE 0 WASTE			
As	0.0022	13	0.0030	0.010	5.8	0.50	500	1.0	500	4.0	2 000				
B	1.1	NA	0.220	0.50	150	25	15 000	50	15 000	200	60 000				
Ba	0.32	716	0.030	0.70	62.5	35	6 250	70	6 250	280	25 000				
Cd	0.0011	<0.020	0.0020	0.003	7.5	0.15	260	0.30	260	1.2	1 040				
Co	0.0011	16	0.0020	0.50	50	25	5 000	50	5 000	200	20 000				
Cr	0.028	113	0.040	0.10	46 000	5.0	800 000	10	800 000	40	██████████				
Cr(VI)	0.028	NA	0.010	0.050	6.5	2.5	500	5.0	500	20	2 000				
Cu	0.0020	59	0.0040	2.0	16	100	19 500	200	19 500	800	78 000				
Hg	0.0011	<3.0	0.00010	0.006	0.93	0.30	160	0.60	160	2.40	640				
Mn	0.0020	488	0.060	0.50	1 000	25	25 000	50	25 000	200	100 000				
Mo	0.17	5.2	0.020	0.070	40	3.5	1 000	7.0	1 000	28	4 000				
Ni	0.0035	51	0.0070	0.070	91	3.5	10 600	7.0	10 600	28	42 400				
Pb	0.0020	41	0.0040	0.010	20	0.50	1 900	1.0	1 900	4.0	7 600				
Sb	0.00022	0.89	0.0070	0.020	10	1.00	75	2.00	75	8.00	300				
Se	0.0089	<2.0	0.0040	0.010	10	0.50	50	1.0	50	4.0	200				
V	0.020	68	0.0030	0.20	150	10	2 680	20	2 680	80	10 720				
Zn	0.0050	314	0.080	5.0	240	250	160 000	500	160 000	2 000	640 000				
TDS	806	██████████	21	1 100	██████████	12 500	██████████	25 000	██████████	100 000	██████████				
Chloride	91	██████████	0.50	300	██████████	15 000	██████████	30 000	██████████	120 000	██████████				
Sulphate as SO ₄	357	██████████	0.40	250	██████████	12 500	██████████	25 000	██████████	100 000	██████████				
NO ₃ as N	3.0	██████████	0.40	11	██████████	550	██████████	1 100	██████████	4 400	██████████				
Fluoride	0.69	NA	0.30	0.15	100	75	10 000	150	10 000	600	40 000				
NA	Not analysed														
N/A	Not available														
	LC ≤ LCT0 and TC ≤ TCT0: Type 4 wastes														
	LCT0 < LV ≤ LCT1 and TC ≤ TCT1: Type 3 Wastes														
	LCT1 < LC ≤ LCT2 and TC ≤ TCT1: Type 2 Waste														
	LCT2 < LC ≤ LCT3 or TCT1 < TC ≤ TCT2: Type 1 Wastes														
	LC > LCT3 or TC > TCT2: Type 0														



Table 4-5: Ash Seepage Water Concentrations versus LCT and TCT values

Camden Power Station Ash: Seepage Water																	
Chemical Species	Seepage water (LC)	Total Concentration (TC)	Limit of Report for LC	LCT0		LCT1		LCT2		LCT3		TCT0		TCT1		TCT2	
				mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg
As	0.0049	NA	0.0030	0.010	5.8	0.50	500	1.0	500	4.0	2 000						
B	2.50	NA	0.220	0.50	150	25	15 000	50	15 000	200	60 000						
Ba	0.063	NA	0.030	0.70	62.5	35	6 250	70	6 250	280	25 000						
Cd	<0.002	NA	0.0020	0.003	7.5	0.15	260	0.30	260	1.2	1 040						
Co	<0.002	NA	0.0020	0.50	50	25	5 000	50	5 000	200	20 000						
Cr	0.0051	NA	0.0030	0.10	46000	5.0	800 000	10	800 000	40							
Cr(VI)	<0.01	NA	0.010	0.050	6.5	2.5	500	5.0	500	20	2 000						
Cu	<0.004	NA	0.0040	2.0	16	100	19 500	200	19 500	800	78 000						
Hg	0.00042	NA	0.00010	0.006	0.93	0.3	160	0.6	160	2.4	640						
Mn	<0.003	NA	0.0030	0.50	1 000	25	25 000	50	25 000	200	100 000						
Mo	0.19	NA	0.020	0.070	40	3.5	1 000	7.0	1 000	28	4 000						
Ni	<0.007	NA	0.0070	0.070	91	3.5	10 600	7.0	10 600	28	42 400						
Pb	<0.004	NA	0.0040	0.010	20	0.50	1 900	1.0	1 900	4.0	7 600						
Sb	NA	NA	0.0070	0.02	10	1.00	75	2.0	75	8.00	300						
Se	0.0047	NA	0.0040	0.010	10	0.50	50	1.0	50	4.0	200						
V	<0.001	NA	0.001	0.20	150	10	2 680	20	2 680	80	10 720						
Zn	<0.01	NA	0.01	5.0	240	250	160 000	500	160 000	2 000	640 000						
TDS	764		21	1 000		12 500		25 000	N/A	100 000	N/A						
Chloride	160		0.50	300		15 000		30 000	N/A	120 000	N/A						
Sulphate as SO ₄	450		0.40	250		12 500		25 000	N/A	100 000	N/A						
NO ₃ as N	<0.1		0.10	11		550		1 100	N/A	4 400	N/A						
Fluoride	<0.05	NA	0.30	1.5	100	75	10 000	150	10 000	600	40 000						
NA	Not analysed																
N/A	Not available																
	LC ≤ LCT0 and TC ≤ TCT0: Type 4 wastes																
	LCT0 < LV ≤ LCT1 and TC ≤ TCT1: Type 3 Wastes																
	LCT1 < LC ≤ LCT2 and TC ≤ TCT1: Type 2 Wastes																
	LCT2 < LC ≤ LCT3 or TCT1 < TC ≤ TCT2 : Type 1 Wastes																
	LC > LCT3 or TC > TCT2: Type 0 Wastes																



Table 4-6: Concentrations of Constituents of the RO Plant Brine versus LCT and TCT values

Camden Power Station Ash: Brine: 70% Recovery Rate															
Chemical Species	Brine from RO Plant(LC)	Total Concentration (TC)	Limit of Report for LC	LCT0	TCT0	LCT1	TCT1	LCT2	TCT1	LCT3	TCT2				
	mg/ℓ	mg/kg	mg/ℓ	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg	mg/ℓ	mg/kg				
As	N/A	NA	N/A	0.010	5.8	0.50	500	1.0	500	4.0	2 000				
B	N/A	NA	N/A	0.50	150	25	15 000	50	15 000	200	60 000				
Ba	0.0250	NA	N/A	0.70	62.5	35	6 250	70	6 250	280	25 000				
Cd	<0.0050	NA	N/A	0.003	7.5	0.15	260	0.30	260	1.2	1 040				
Co	<0.0050	NA	N/A	0.50	50	25	5 000	50	5 000	200	20 000				
Cr	0.10	NA	N/A	0.10	46 000	5.0	800 000	10	800 000	40					
Cr(VI)	N/A	NA	N/A	0.050	6.5	2.5	500	5.0	500	20	2 000				
Cu	<0.0050	NA	N/A	2.0	16	100	19 500	200	19 500	800	78 000				
Hg	0.0040	NA	N/A	0.006	0.93	0.03	160	0.6	160	2.4	640				
Mn	<0.005	NA	N/A	0.50	1 000	25	25 000	50	25 000	200	100 000				
Mo	0.10	NA	N/A	0.070	40	3.5	1 000	7.0	1 000	28	4 000				
Ni	<0.0050	NA	N/A	0.070	91	3.5	10 600	7.0	10 600	28	42 400				
Pb	0.27	NA	N/A	0.010	20	0.50	1 900	1.0	1 900	4.0	7 600				
Sb	N/A	NA	N/A	0.02	10	1.00	75	2.00	75	8.00	300				
Se	N/A	NA	N/A	0.010	10	0.50	50	1.0	50	4.0	200				
V	0.10	NA	N/A	0.20	150	10	2 680	20	2 680	80	10 720				
Zn	<0.0050	NA	N/A	5.0	240	250	160 000	500	160 000	2 000	640 000				
TDS	2 150*		N/A	1 000		12 500		25 000	N/A	100 000	N/A				
Chloride	380		N/A	300		15 000		30 000	N/A	120 000	N/A				
Sulphate as SO ₄	2 080		N/A	250		12 500		25 000	N/A	100 000	N/A				
NO ₃ as N	3.32		N/A	11		550		1 100	N/A	4 400	N/A				
Fluoride	3.47	NA	N/A	1.5	100	75	10 000	150	10 000	600	40 000				
NA	Not analysed														
N/A	Not available														
	Values in red are theoretical														
*	Calculated TDS at 60% clean water recovery rate using a µS/cm to mg/ℓ conversion factor of 0.65														
	LC ≤ LCT0 and TC ≤ TCT0: Type 4 wastes														
	LCT0 < LV ≤ LCT1 and TC ≤ TCT1: Type 3 Wastes														
	LCT1 < LC ≤ LCT2 and TC ≤ TCT1: Type 2 Waste														
	LCT2 < LC ≤ LCT3 or TCT1 < TC ≤ TCT2: Type 1 Wastes														
	LC > LCT3 or TC > TCT2: Type 0														



5. DISCUSSION AND CONCLUSIONS

In terms of the DEA's National Norms and Standards, the Camden wet ash was subjected to a TC extract and a DI water leach. Two samples were used in the assessment, namely dusting ash (fine ash) and ashing ash (course) ash. In addition, the water leaching from the base of the existing ash disposal facility was also analysed and compared to the respective LCT values. The seepage water was therefore also classified in terms of the National Norms and Standards, as it is seen as the actual risk posed by the ash disposal facility to the receiving environment.

The DI water leach scenario is applicable in the case that ash is mono-disposed or stored in the environment at a permanent storage facility, i.e., the waste is stored for longer than 90 days. Based on the DI water leach results, and taking the concentrations of the water fractions of the wet ash samples into account, both the dusting and ashing ash samples are classified as Type 3 wastes requiring disposal on a landfill with a Class C barrier system – see **Figure 5-1**.

This barrier system is considered appropriate for the wet ash disposal facility provided the drainage layer on top of the barrier system contains drainage pipes of adequate size, spacing and strength to ensure atmospheric pressure within the drainage application for the service life of the ash disposal facility (DEA, 2013b). However, in the case of the wet ash, the DWA may require that a lagoon barrier design as per the DWAF's Minimum Requirements be installed – see **Figure 5-2**. In addition, it should be noted that the National Norms and Standards require that the disposal of liquid waste must be phased out over a period of six years from the date that the National Norms and Standards were promulgated. If the authorities insist on this approach, it may have significant cost implications for the Camden Power Station. Therefore it is recommended that agreement be reached with the authorities on the long term management scenario of the ash disposal facility prior to the barrier system being designed.

The RO plant brine is also classified as a Type 3 waste. In the case that the brine is co-disposed with the ash on the new ash disposal facility, a Class C landfill barrier is considered appropriate for the wet ash and brine disposal facility. As with the wet ash only disposal scenario, it is a requirement that liquid waste should be disposed of in hazardous lagoon facilities, but provided the drainage layer on top of the Class C barrier system contains drainage pipes of adequate size, spacing and strength to ensure atmospheric pressure within the drainage application for the service life of the ash disposal facility, the co-disposal scenario is considered appropriate. It has been shown that ash has significant capacity to adsorb salts, which is also the case at Camden. The TDS of the ashing water (average of the dusting and ashing ash values is 1 424 mg/l) has a significantly higher TDS value than that of the seepage water (764 mg/l) – see **Table 4-1**, **Table 4-3** and **Table 4-5**. The co-disposal of the brine with the wet ash may therefore be regarded as treatment of the RO plant brine.

In the case that the RO plant brine is mono-disposal, the barrier design will have to comply with the performance specifications of that of a hazardous waste lagoon.

It is important to note that the disposal of brines or wastes with a high salt content (TDS > 5%) and a leachable concentration for TDS of more than 100 000 mg/l needs to be phased out within eight (8) years from the date of promulgation of the National Norms and Standards (DEA, 2013b). However, the brine from the RO plant at Camden has a TDS of only 2 150 mg/l (0.215%). Therefore the requirement of phasing out the disposal of the Camden RO brine is not applicable as the TDS is lower than 5%.

Table 5-1 below summarises the classification of the wet ash and RO plant brine and also indicates the recommended barrier systems for the various disposal scenarios.



Table 5-1: Waste Type and Recommended Class of Landfill Required

Waste	Type of Waste	Disposal Scenario	Class of Landfill	Recommended Barrier System
Ash + Ash Carrier Water	Type 3: Low Risk Waste	Mono-disposal	Class C	Class C ⁽¹⁾
Brine from RO Plant	Type 3: Low Risk Waste	Mono-disposal	H:H Lagoon	H:H Lagoon
Ash + Ash Carrier Water + RO Plant Brine	Type 3: Low Risk Waste	Co-disposal	Class C	Class C ⁽¹⁾

(1): Provided the drainage layer on top of the barrier system contains drainage pipes of adequate size, spacing and strength to ensure atmospheric pressure within the drainage application for the service life of the ash disposal facility

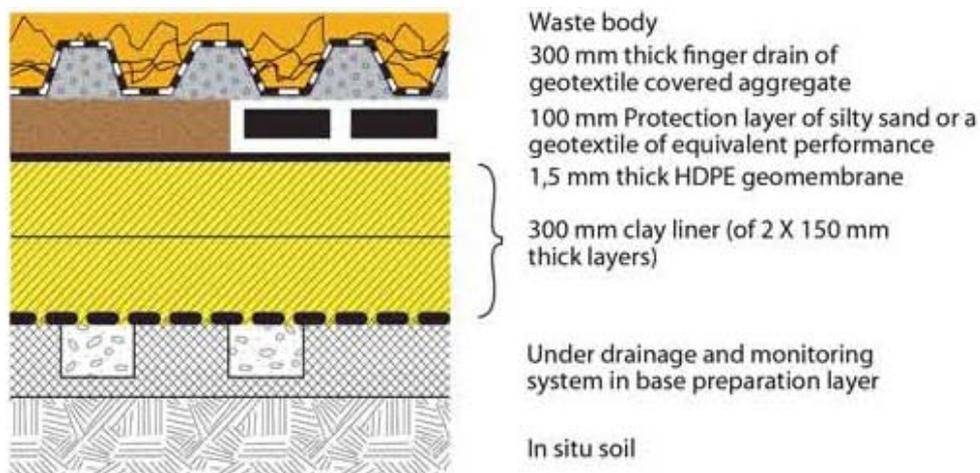


Figure 5-1: Class C landfill barrier system (DEA, 2013b)

Hazardous Waste Lagoons

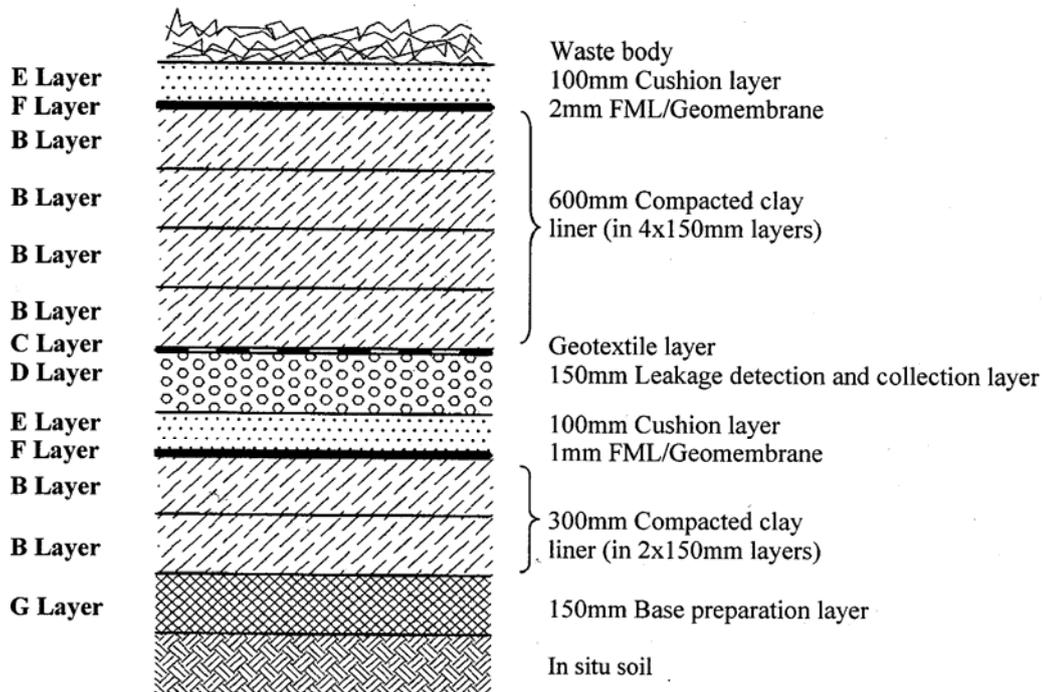


Figure 5-2: H:H Lagoon barrier system (DWAf, 1998b)

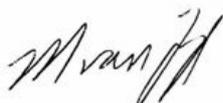
6. RECOMMENDATIONS

The following recommendations are made:

- The intended barrier design of the new wet ash disposal facility for Camden Power Station should be presented, discussed and agreed upon with the Department of Water Affairs prior to the design being developed;
- A Class 3 barrier design, which is the recommended barrier system by J&W, for the new wet ash disposal facility should incorporate a drainage layer on top of the barrier system containing drainage pipes of adequate size, spacing and strength to ensure atmospheric pressure within the drainage application for the service life of the ash disposal facility as per the DEA National Norms and Standards or as agreed with the Department of Water Affairs.
- If the RO brine is to be mono disposed, the barrier system for the disposal facility must comply with the performance requirements of a Hazardous Lagoon as specified in the Department of Water Affairs' Minimum Requirements of 1998 due to the head of water on the barrier system.
- If Eskom decides to co-dispose the RO plant brine with the wet ash, it should be motivated that the co-disposal is considered treatment of the brine. The brine from the Camden RO plant does not qualify as a brine in terms of the specification given in the National Norm and Standards.

7. REFERENCES

- i. Department of Water Affairs and Forestry, 1998a. Minimum Requirements for the Handling, Classification and Disposal of Hazardous Waste, Second Edition. Department of Water Affairs, Pretoria.
- ii. Department of Water Affairs and Forestry, 1998b. Minimum Requirements for Waste Disposal by Landfill, Second Edition. Department of Water Affairs, Pretoria.
- iii. Department of Environmental Affairs and Tourism, 2008. Waste delisting procedure, April 2008. Department of Environmental Affairs and Tourism, Pretoria.
- iv. Department of Environmental Affairs, 2013a. National norms and standards for the assessment of waste for landfill disposal. R635 of 23 August 2013, Government Gazette 36784 of 23 August 2013, Government Printer, Pretoria.
- v. Department of Environmental Affairs, 2013b. National norms and standards for disposal of waste to landfill. R636 of 23 August 2013, Government Gazette 36784 of 23 August 2013, Government Printer, Pretoria.
- vi. Legge, K., 2011. *Verbal communication*. Department of Water Affairs.
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Tolmay Hopkins
Reviewer



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17 September 2014

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ZITHOLELE CONSULTING (PTY) LTD

WASTE CLASSIFICATION OF POWER STATION
ASH AND BRINE FROM THE CAMDEN POWER STATION

Report: JW164/11/D116 - REV 6

Appendix A

SGS SOUTH AFRICA: LABORATORY CERTIFICATES





TEST REPORT

SGS South Africa (Pty) Ltd.
58 Melville Street
Booyens
Johannesburg

Sarah Newton
SGS Environmental Services
259 Kent Avenue
Randburg

MINERALOGICAL REPORT No: MIN 0911/192

Work Requested By: Sarah Newton
On Behalf Of: SGS Environmental
Date issued: 05 October 2011
Investigator: O.D Mosinyi

Analysis of Sample 1881-001 by XRD

O.D. Mosinyi

Mineralogist

L.L. Coetzee

Manager: Mineralogy

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1. INTRODUCTION

Sarah Newton, on behalf of SGS Environmental Services, submitted one sample for X-ray diffraction mineralogical examination. The sample was labelled 1881-001, a dry ash sample.

2. METHODOLOGY

The sample was pulverized and analysed by X-ray diffraction utilising a Panalytical X'pert Pro Diffractometer employing Co-K α radiation. Data interpretation was by means of Panalytical Highscore Plus analytical software, in conjunction with the PDF2 database. The XRD analysis was used to identify and quantify the crystalline phases present in the sample.

3. RESULTS

3.1 X-ray Diffraction Analyses

The crystalline phases that were detected by XRD are listed below in Table 1, and the diffractogram for the sample is shown in figure 1. There were four crystalline phases that were detected by XRD. These were mullite which made up 45.2% of the sample, and quartz which also accounted for 45.2 % of the sample, calcite accounted for 6.5 % of the sample and lastly magnetite accounted for 3.1 % of the sample.

Table 1: Crystalline phases as determined by X-ray Diffraction

Mineral	Approx. Formula	01881-001 Mass %
Mullite	$\text{Al}_6\text{Si}_2\text{O}_{13}$	45.2
Quartz	SiO_2	45.2
Calcite	CaCO_3	6.5
Magnetite	Fe_3O_4	3.1

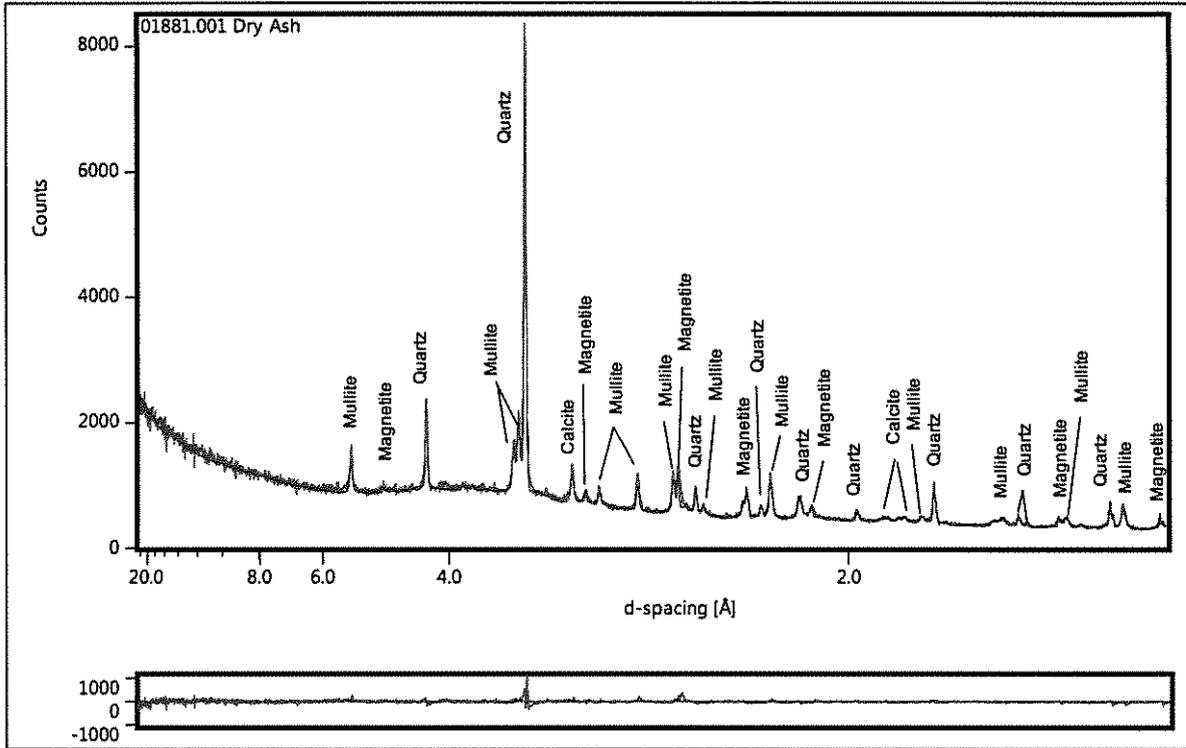


Figure 1: X-ray Diffractogram showing the composition of the sample 1881-001. The diffractogram in red shows the measured pattern, while the blue shows the calculated pattern obtained as part of the Rietveld refinement. The lower red pattern shows the difference between the measured and calculated pattern.



ANALYTICAL REPORT

CLIENT DETAILS

Contact **Marius Van Zyl**
Client **Jones & Wagener (Pty) Ltd**
Address **P.O. Box 1434
Rivonia
2128**

Telephone **011 519 0200**
Facsimile **011 519 0201**
Email **vanzyl@jaws.co.za**
Project **11521199**
Order Number **DI66/MVZ/19829**
Samples **1**
Sample matrix **SOIL**

LABORATORY DETAILS

Laboratory **SGS South Africa (Pty) Limited**
Address **259 Kent Avenue
Ferndale, 2194**
Telephone **+27 (0)11 781 5689**

Laboratory Manager **Mark Baird (acting)**
SGS Reference **JB11-01871 R0**
Report Number **0000001521**
Date Received **2011/09/12 11:49:42AM**
Date Reported **2011/09/30 09:33:06AM**

COMMENTS

The document is issued in accordance with SANAS's accreditation requirements.
Accredited for compliance with ISO/IEC 17025. SANAS accredited laboratory T0107.



Filter cake samples not dried prior to testing.

Sample(s) leached using ARLP leachate. Results reported on leachate.

SIGNATORIES

Gladness Radebe
Technical Supervisor/Technical Signatory

Sarah Newton
Technical Consultant/Technical Signatory



ANALYTICAL REPORT

JB11-01871 R0

Report number 0000001521
Client reference: 11521199

Sample Number JB11-01871.001
Sample Name Dusting Ash

Parameter	Units	LOR	
Acid Rain Leaching Procedure (ARLP) Method:			
Final pH*	-	-	7.9

Conductivity - Water Method: ME-ANA-AN-007			
Conductivity	mS/m	2.0	120

Total Dissolved Solids (TDS) in water Method: ME-ANA-AN-011			
Total Dissolved Solids	mg/l	21.0	528

Anions by Ion Chromatography Method: ME-ANA-AN-AN014			
Fluoride	mg/l	0.050	<0.050
Chloride	mg/l	0.050	2.5
Nitrate	mg/l	0.10	15
Sulphate	mg/l	0.050	180

Hexavalent Chromium by UV-VIS Method: ME-ANA-AN-018			
Hexavalent Chromium*	mg/l	0.010	0.40

Ammonia as N by UV Method: APHA4500_NH3			
Ammonia*	mg/l	0.050	<0.050

ICP-OES Metals in Water (Dissolved) Method: ME-ANA-AN-027			
Silver	mg/l	0.0020	<0.0020
Aluminium	mg/l	0.020	0.069
Boron	mg/l	0.0050	2.3
Barium	mg/l	0.0020	0.21
Beryllium	mg/l	0.00010	<0.00010
Calcium	mg/l	0.50	200
Iron	mg/l	0.050	<0.050
Potassium	mg/l	0.20	1.4
Lithium	mg/l	0.0050	0.073
Magnesium	mg/l	0.010	45
Sodium	mg/l	0.50	5.4
Silicon	mg/l	1.0	11
Strontium	mg/l	0.0010	2.6
Titanium	mg/l	0.0050	0.023
Vanadium	mg/l	0.0010	0.38
Zinc	mg/l	0.010	<0.010

ICP-MS Metals (Dissolved) Method: ME-ANA-AN-026			
Arsenic	mg/l	0.0030	0.080
Bismuth	mg/l	0.0010	<0.0010
Cadmium	mg/l	0.0020	<0.0020
Cobalt	mg/l	0.0020	<0.0020
Chromium	mg/l	0.0030	0.40
Copper	mg/l	0.0040	<0.0040
Mercury	mg/l	0.00010	0.0020
Manganese	mg/l	0.0030	0.049
Molybdenum	mg/l	0.0070	0.14
Nickel	mg/l	0.0070	0.014
Lead	mg/l	0.0040	<0.0040
Antimony	mg/l	0.0070	0.013
Selenium	mg/l	0.0040	0.026
Tin	mg/l	0.0070	<0.0070

METHOD

METHODOLOGY SUMMARY

FOOTNOTES

IS	Insufficient sample for analysis.	QFH	QC result is above the upper tolerance
LNR	Sample listed, but not received.	QFL	QC result is below the lower tolerance
*	This analysis is not covered by the scope of accreditation.	-	The sample was not analysed for this analyte
^	Performed by outside laboratory.		
LOR	Limit of Reporting		
↑↓	Raised or Lowered Limit of Reporting		

Samples analysed as received.
Solid samples expressed on a dry weight basis.

Unless otherwise indicated, samples were received in containers fit for purpose.

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ANALYTICAL REPORT

CLIENT DETAILS

Contact **Marius Van Zyl**
Client **Jones & Wagener (Pty) Ltd**
Address **P.O. Box 1434
Rivonia
2128**

Telephone **011 519 0200**
Facsimile **011 519 0201**
Email **vanzyl@jaws.co.za**
Project **11521195**
Order Number **DI66/MVZ/19829**
Samples **3**
Sample matrix **WATER**

LABORATORY DETAILS

Laboratory **SGS South Africa (Pty) Limited**
Address **259 Kent Avenue
Fermdale, 2194**
Telephone **+27 (0)11 781 5689**

Laboratory Manager **Mark Baird (acting)**
SGS Reference **JB11-01869 R0**
Report Number **0000001519**
Date Received **2011/09/12 10:00:46AM**
Date Reported **2011/09/30 09:26:12AM**

COMMENTS

The document is issued in accordance with SANAS's accreditation requirements.
Accredited for compliance with ISO/IEC 17025. SANAS accredited laboratory T0107.



Samples filtered prior to analysis.

SIGNATORIES

Gladness Radebe
Technical Supervisor/Technical Signatory

Sarah Newton
Technical Consultant/Technical Signatory



ANALYTICAL REPORT

JB11-01869 R0

Report number: **000001519**
 Client reference: **11521195**

Sample Number	JB11-01869.001	JB11-01869.002	JB11-01869.003
Sample Name	Seepage Water	Ashing Water	Dusting Water

Parameter	Units	LOR			
pH in water Method: ME-ANA-AN-016					
pH		0.10	8.4	11.4	12.2

Conductivity - Water Method: ME-ANA-AN-007					
Conductivity	mS/m	2.0	160	190	740

Total Dissolved Solids (TDS) in water Method: ME-ANA-AN-011					
Total Dissolved Solids	mg/l	21.0	764	856	1992

Anions by ion Chromatography Method: ME-ANA-AN-AN014					
Fluoride	mg/l	0.050	<0.050	0.74	0.73
Chloride	mg/l	0.050	160	97	120
Nitrate	mg/l	0.10	<0.10	3.2	0.64
Sulphate	mg/l	0.050	450	380	210

Ammonia as N by UV Method: APHA4500_NH3					
Ammonia*	mg/l	0.050	<0.050	<0.050	0.066

Hexavalent Chromium by UV-VIS Method: ME-ANA-AN-018					
Hexavalent Chromium*	mg/l	0.010	<0.010	0.030	0.15

ICP-OES Metals in Water (Dissolved) Method: ME-ANA-AN-027					
Silver	mg/l	0.0020	0.0037	0.0041	0.026
Aluminium	mg/l	0.020	<0.020	1.2	0.19
Boron	mg/l	0.0050	2.5	1.1	0.11
Barium	mg/l	0.0020	0.063	0.34	1.3
Beryllium	mg/l	0.00010	-1.30551E-	-2.85557E-	-6.56818E-
Calcium	mg/l	0.50	110	190	760
Iron	mg/l	0.050	<0.050	<0.050	<0.050
Potassium	mg/l	0.20	39	27	88
Lithium	mg/l	0.0050	0.61	0.85	3.8
Magnesium	mg/l	0.010	8.7	0.072	<0.010
Sodium	mg/l	0.50	240	160	210
Silicon	mg/l	1.0	1.7	7.6	<1.0
Strontium	mg/l	0.0010	3.9	3.6	39
Titanium	mg/l	0.0050	<0.0050	<0.0050	0.0098
Vanadium	mg/l	0.0010	<0.0010	0.020	0.0021
Zinc	mg/l	0.010	<0.010	<0.010	<0.010

ICP-MS Metals (Dissolved) Method: ME-ANA-AN-026					
Arsenic	mg/l	0.0030	0.0049	<0.0030	<0.0030
Bismuth	mg/l	0.0010	<0.0010	<0.0010	<0.0010
Cadmium	mg/l	0.0020	<0.0020	<0.0020	<0.0020
Cobalt	mg/l	0.0020	<0.0020	<0.0020	<0.0020
Chromium	mg/l	0.0030	0.0051	0.029	0.15
Copper	mg/l	0.0040	<0.0040	<0.0040	<0.0040
Mercury	mg/l	0.00010	0.00042	0.0012	<0.00010
Manganese	mg/l	0.0030	<0.0030	<0.0030	<0.0030
Molybdenum	mg/l	0.0070	0.19	0.18	0.19
Nickel	mg/l	0.0070	<0.0070	<0.0070	<0.0070
Lead	mg/l	0.0040	<0.0040	<0.0040	<0.0040
Selenium	mg/l	0.0040	0.0047	0.0094	<0.0040
Tin	mg/l	0.0070	<0.0070	<0.0070	<0.0070



METHOD SUMMARY

JB11-01869 R0

Report number: 000001519
Client reference: 11521195

METHOD

METHODOLOGY SUMMARY

FOOTNOTES

IS	Insufficient sample for analysis.	QFH	QC result is above the upper tolerance
LNR	Sample listed, but not received.	QFL	QC result is below the lower tolerance
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^	Performed by outside laboratory.		
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Samples analysed as received.
Solid samples expressed on a dry weight basis.

Unless otherwise indicated, samples were received in containers fit for purpose.

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T0107



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Sarah Newton

SGS Environmental Services SA

P.O. Box 82582

Southdale

2135

TEST REPORT

Lab Ref LA117646
Client Ref **JB11 - 01881**
Project DEFAULT
Product Code SOLIDS

Status Final
Received 14/09/11
Reported 10/10/11

Samples 2
First Sample 1881 - 001
Last Sample WASTE ROCK
Pages 10

Notes

Empty box for notes.

Technical Signatory Name: **Signature:**.....

Technical Signatory Name: **Signature:**.....

Technical Signatory Name: **Signature:**.....

On behalf of: SGS South Africa

The results in the following analytical report pertain to this laboratory for preparation and/or analysis as requested by SGS Environmental Services SA.

The analytical results reported herein refer to the samples as received and are based on a dry basis where applicable.

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Lab Ref LA117646
 Client Ref **JB11 - 01881**
 Project DEFAULT
 Reported 10/10/11
 Status Final
 Page Page 2 of 10

TEST REPORT

Scheme	WtRec	Al	Ba	Ca	Cr	Cu
	WGH79	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B
Units	g	%	ppm	%	ppm	ppm
Detection Limit	0.01	0.01	5	0.01	1	0.5
1881 - 001	34.50	10.5	716	3.50	113	59.4
WASTE ROCK	-	0.28	94	0.03	22	14.6
GEOSTATS		4.34	36	1.13	1750	3880
LKSD-3SA		5.67	638	1.49	-	-
OREAS 100A		5.58	417	1.05	39	183
OREAS 101A		5.78	180	1.23	39	-
BLANK		<0.01	<5	<0.01	<1	<0.5
1881 - 001		10.8	777	3.63	119	62.4

- not analysed / -- element not determined / I.S. insufficient sample / L.N.R. listed not received / U.T.D. Unable To Determine

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TEST REPORT

Scheme	Fe	K	Li	Mg	Mn	Na
	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B
Units	%	%	ppm	%	ppm	%
Detection Limit	0.01	0.01	1	0.01	5	0.01
1881 - 001	6.86	0.50	181	0.82	488	0.12
WASTE ROCK	0.72	0.08	<1	<0.01	128	0.02
GEOSTATS	4.75	3.41	9	0.52	5230	1.60
LKSD-3SA	4.01	2.02	27	1.14	1410	1.97
OREAS 100A	4.21	3.79	20	0.85	579	0.14
OREAS 101A	10.4	2.26	44	1.24	1020	0.08
BLANK	<0.01	<0.01	<1	<0.01	<5	<0.01
1881 - 001	7.03	0.52	188	0.86	508	0.12

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Page Page 4 of 10

TEST REPORT

	P	S	Sr	Ti	V	Zn
Scheme	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B
Units	ppm	%	ppm	%	ppm	ppm
Detection Limit	50	0.01	0.5	0.01	1	1
1881 - 001	1130	0.20	1010	0.71	68	314
WASTE ROCK	210	0.04	<0.5	0.01	3	39
GEOSTATS	460	0.96	43.7	0.21	45	5230
LKSD-3SA	1110	-	237	-	-	-
OREAS 100A	510	0.06	22.5	-	-	41
OREAS 101A	-	0.13	10.0	-	-	101
BLANK	<50	<0.01	<0.5	<0.01	<1	5
1881 - 001	1190	0.22	1050	0.74	77	336

- not analysed / -- element not determined / I.S. insufficient sample / L.N.R. listed not received / U.T.D. Unable To Determine

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 Page Page 5 of 10

TEST REPORT

Scheme	Zr	Ag	As	Be	Bi	Cd
	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B
Units	ppm	ppm	ppm	ppm	ppm	ppm
Detection Limit	0.5	0.02	1	0.1	0.04	0.02
1881 - 001	254	<0.02	13	5.6	1.24	<0.02
WASTE ROCK	54.2	<0.02	2	0.1	0.31	<0.02
GEOSTATS	68.2	48.0	13	-	-	-
LKSD-3SA	-	2.87	27	1.8	-	-
OREAS 100A	121	-	-	-	-	-
OREAS 101A	91.0	-	-	-	-	-
BLANK	<0.5	<0.02	<1	<0.1	<0.04	<0.02
1881 - 001	275					
1881 - 001		<0.02	13	5.8	1.25	<0.02

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Page Page 6 of 10

TEST REPORT

	Co	Mo	Ni	Pb	Sb	Se
	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B	ICM40B
Units	ppm	ppm	ppm	ppm	ppm	ppm
Detection Limit	0.1	0.05	0.5	0.5	0.05	2
1881 - 001	16.4	5.18	51.3	41.4	0.89	<2
WASTE ROCK	1.6	3.71	5.3	7.6	0.17	<2
GEOSTATS	2070	-	4030	1.21%	11.3	-
LKSD-3SA	29.0	-	46.7	29.3	1.36	-
OREAS 100A	16.4	20.7	-	13.4	-	-
OREAS 101A	47.0	20.5	-	21.3	-	-
BLANK	<0.1	<0.05	<0.5	<0.5	0.09	<2
1881 - 001	16.6	5.22	52.0	41.7	0.90	<2

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TEST REPORT

Scheme	Sn	Hg	Si
	ICM40B	IMS12B	ICP90A
Units	ppm	ppm	%
Detection Limit	0.3	3	0.1
1881 - 001	4.4	<3	19.2
WASTE ROCK	0.5	<3	20.8
GEOSTATS	-		
LKSD-3SA	-		
OREAS 100A	-		
OREAS 101A	-		
BLANK	<0.3		
BLANK		<3	
SARM5			-
BLANK			<0.1
1881 - 001			19.9
BCS176/2			1.27
1881 - 001		<3	
CCU-1C		30	
GXR-1		4	
1881 - 001	4.4		

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Client Ref **JB11 - 01881**
Project DEFAULT
Reported 10/10/11
Status Final
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TEST REPORT**APPENDIX A - METHODS**

METHOD NUMBER	METHOD DESCRIPTION	SCHEME CODE
ME-ZA-[MINANA]-[BYZ(FAS)]AN-001	Au by Lead Fusion followed by Atomic Absorption analysis or Gravimetric analysis	FAALA01, FAALA01D, FAGLA01, FAGLA02, FAGLA03, FAGLA04, FAGLA05
ME-ZA-[MINANA]-[BYZ(FAS)]AN-002	Au, Pt, Pd by Lead Fusion followed by	FAI313
ME-ZA-[MINANA]-[BYZ(FAS)]AN-003	Pt, Pd, Rh, Ru, Ir by Nickel Sulphide, ICP-OES finish	FAI363
ME-ZA-[MINANA]-[BYZ(XRF)]AN-001	Major Element Oxides by Borate fusion XRF	XRF79V, XRF79C
ME-ZA-[MINANA]-[BYZ(XRF)]AN-002	Base Metals by Potassium Pyrosulphate Fusion XRF	XRF77R
ME-ZA-[MINANA]-[BYZ(AAS)]AN-001	Acid Soluble Cu and Ni by Acid digestion and analysis by AAS	AAS13C
ME-ZA-[MINANA]-[BYZ(LEC)]AN-001	Total Sulphur and Carbon by Leco Combustion Infrared Detection	CSALA01, CSALA06
ME-ZA-[MINANA]-[BYZ(ICM)]AN-001	Total & Dissolved metals by ICP-OES & ICP-MS	ICP84T & IMS84T
ME-ZA-[MINANA]-[BYZ(XRF)]AN-003	Uranium Oxide, pressed powder analysis using XRF spectrometer	XRF75G
ME-ZA-[MINANA]-[BYZ(FAS)]AN-005	Rh by Pd fusion by ICP-OES finish	FAI353
ME-ZA-[MINANA]-[BYZ(WET)]AN-001	Chloride by Potentiometric titration	CLA27V

- not analysed / -- element not determined / I.S. insufficient sample / L.N.R. listed not received / U.T.D. Unable To Determine

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Client Ref JB11 - 01881
Project DEFAULT
Reported 10/10/11
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Page Page 9 of 10

TEST REPORT**METHOD DESCRIPTION**

Silver (Ag) by Fire Assay, gravimetric finish
Trace elements by pressed pellet, XRF
Sulphide Sulphur (S₂-) by Leco
Elemental sulphur (S⁰) by gravimetric finish
Aqueous sulphate (SO₄) by Dionex
Sulphate (SO₄) on solids by Dionex
Carbonate (CO₃) by LECO
Graphite carbon by LECO
Organic carbon by LECO
pH determination
Conductivity (EC) determination
Total Hardness as CaCO₃ (calc from ICP Ca, Mg analyses)
Anions by IC (F, Cl, NO₂, NO₃, SO₄)
Ammonia (NH₃) by spectroquant
Phosphate (PO₄) by colourmetric analysis
Chemical Oxygen Demand (COD) by spectroquant
Suspended solids (TSS)
Total dissolved solids (TDS), gravimetric finish (180 °C)/Electrometric, conductivity meter
Alkalinity by titration
Chloride (Cl) by titration (solutions)
Chloride (Cl) by titration (solids)
Fluoride (F) by ISE (solutions)
Fluoride (F) by ISE (solids)
Acid Base Accounting (ABA)
Net acid generation (NAG) test (incl. S species)
Short term leach testing (ARLP, TCLP, SPLP, etc)
Deionised water (DI) leach (2 hours, L:S=10)
Cyanide (CN) species - Free, WAD & Total
Thiocyanate (SCN) by IC
Metals by AAS (solutions)
Gold (Au) in CN solutions by AAS
Silver (Ag) by acid digestion, AAS
Arsenic (As) by Aqua Regia digestion, AAS
Multi Acid digestion, AAS finish
Acid soluble Cu, Co by Sulphuric Acid leach, AAS
Aqua Regia digestion, ICP-OES finish
Multi Acid digestion, ICP-OES finish
Sodium Peroxide fusion, ICP-OES finish

SCHEME CODE

FAGLA02
XRF75G
CSA08V
CSA12V
CLA31V
CSA11V
CSA02V
CSA10V
CSA03V
ISE06T
ISE09V
ICP84B
CLA31V
CLA23V
CLA22V
CLA24V
PHY18V
ISE10V
CLA28V
CLA27V
CLA04E
ISE07W
ISELA01
CLA41V
CLA43V
CLA40V
Leach
CLA25V
CLA31V
AAS84T
SOL81T
AAS14E
AAS11C
AAS40D
AAS72C
ICP13E
ICP40D
ICP91B

- not analysed / -- element not determined / I.S. Insufficient sample / L.N.R. listed not received / U.T.D. Unable To Determine

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Internet: www.sgs.com

Lab Ref LA117646
Client Ref JB11 - 01881
Project DEFAULT
Reported 10/10/11
Status Final
Page Page 10 of 10

TEST REPORT**METHOD DESCRIPTION**

Semi quantitative ICP-OES +ICP-MS scan, Aqua Regia digestion
As, Hg, Se, Te by Aqua Regia digestion, ICP-MS finish
Multi Acid digestion, semi quantitative scan, ICP-OES + ICP-MS
Multi acid digestion, ICP-MS
Rare Earth Elements (REE) by Na2O2 fusion, ICP-MS
Free acid titration
Chloride (Cl) by manual titration (Metallurgical)
As 3+ by titration
As 5+ by calculation
Lime (CaO) by titration
Lime (CaO), calculation after AAS analysis
Ferrous (Fe2+) iron by titration (solids)
Ferrous (Fe2+) iron by titration (solutions)
Ferric (Fe3+) iron by diff (incl. Fe total, Fe2+) - solids
Ferric (Fe3+) iron by diff (incl. Fe total, Fe2+) - solutions
Iron (Fe) by titration (solids)
Tin (Sn) by titration (solids)
Zinc (Zn) by EDTA titration (solids)
Hexavalent chromium (Cr6+) in solutions
Manganese (Mn) by back titration
Vanadium (V) by titration
Chrome (Cr) by back titration
Relative Density/Specific Gravity (by Le Chatelier flask)
Bulk density
Relative Density/Specific Gravity (by Helium pycnometer)
Grain density
Moisture (105 °C)
Ash/LOI (1050 °C)

SCHEME CODE

ICM12B
IMS12Q
ICM40B
IMS40B
IMS90A
CLA15F
CLA26V
CLA32V
CLA32V
CLA07C
CLA07C
CLA34V
CLA34V
CLA34V
CLA35V
CLA35V
CON14V
CON12V
CLA21V
CON15V
CON16V
CON10B
PHY04V
PHY21V
PHY03V
PHY20V
PHY08D
PHY01K

- not analysed / -- element not determined / I.S. insufficient sample / L.N.R. listed not received / U.T.D. Unable To Determine

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ANALYTICAL REPORT

CLIENT DETAILS

Contact **Marius Van Zyl**
Client **Jones & Wagener (Pty) Ltd**
Address **P.O. Box 1434
Rivonia
2128**

Telephone **011 519 0200**
Facsimile **011 519 0201**
Email **vanzyl@jaws.co.za**
Project **(Not specified)**
Order Number **DI66/MVZ/19829**
Samples **1**
Sample matrix **SOIL**

LABORATORY DETAILS

Laboratory **SGS South Africa (Pty) Limited**
Address **259 Kent Avenue
Ferndale, 2194**
Telephone **+27 (0)11 781 5689**

Laboratory Manager **Mark Baird (acting)**
SGS Reference **JB11-01881 R0**
Report Number **0000001593**
Date Received **2011/09/13 12:15:20PM**
Date Reported **2011/10/10 11:32:03AM**

COMMENTS

Whilst SGS laboratories conform to ISO/IEC 17025 standards, results of analysis in this report fall outside of the current scope of accreditation.

Testing subcontracted to SGS Booyens.

Mineralogy results contained in their report, MIN 0911/192, appended.

SIGNATORIES

Gladness Radebe
Technical Supervisor/Technical Signatory

Sarah Newton
Technical Consultant/Technical Signatory

ANALYTICAL REPORT

JB11-01881 R0

Report number: 0000001593
Client reference: DI66/MVZ/19829

Sample Number	JB11-01881.001
Sample Name	Dry Ash
Sample Matrix	Soil

Parameter	Units	LOR
SUB_Mineralogy Method: SUB		
XRD scan	No unit	- MIN 0911/192

Parameter	Units	LOR
SUB_SGS Booyens Method: SUB_BOOY		
Silver	ppm	0.020 <0.020
Aluminium	%	0.010 11
Arsenic	ppm	1.0 13
Barium	ppm	5.0 720
Beryllium	ppm	0.10 5.6
Bismuth	ppm	0.040 1.2
Calcium	%	0.010 3.5
Cadmium	ppm	0.020 <0.020
Chromium	ppm	1.0 110
Cobalt	ppm	0.10 16
Copper	ppm	0.50 59
Iron	%	0.010 6.9
Mercury	ppm	3.0 <3.0
Potassium	%	0.010 0.50
Lithium	ppm	1.0 180
Magnesium	%	0.010 0.82
Manganese	ppm	5.0 490
Molybdenum	ppm	0.050 5.2
Sodium	%	0.010 0.12
Nickel	ppm	0.50 51
Phosphorus	ppm	50 1100
Lead	ppm	0.50 41
Sulphur	%	0.010 0.20
Antimony	ppm	0.050 0.89
Selenium	ppm	2.0 <2.0
Silicon	%	0.10 19
Tin	ppm	0.30 4.4
Strontium	ppm	0.50 1000
Titanium	%	0.010 0.71
Vanadium	ppm	1.0 68
Zinc	ppm	1.0 310
Zirconium	ppm	0.50 250

METHOD SUMMARY

JB11-01881 R0

Report number 0000001593
Client reference: D166/MVZ/19829

METHOD ————— METHODOLOGY SUMMARY

FOOTNOTES

IS	Insufficient sample for analysis.	QFH	QC result is above the upper tolerance
LNR	Sample listed, but not received.	QFL	QC result is below the lower tolerance
•	This analysis is not covered by the scope of accreditation.	-	The sample was not analysed for this analyte
^	Performed by outside laboratory.		
LOR	Limit of Reporting		
↑↓	Raised or Lowered Limit of Reporting		

Samples analysed as received.
Solid samples expressed on a dry weight basis.

Unless otherwise indicated, samples were received in containers fit for purpose.

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ANALYTICAL REPORT (Amended)

CLIENT DETAILS

Contact **Marius Van Zyl**
Client **Jones & Wagener (Pty) Ltd**
Address **P.O. Box 1434
Rivonia
2128**

Telephone **011 519 0200**
Facsimile **011 519 0201**
Email **vanzyl@jaws.co.za**
Project **11521198**
Order Number **DI66/MVZ/19829**
Samples **2**
Sample matrix **SOIL**

LABORATORY DETAILS

Laboratory **SGS South Africa (Pty) Limited**
Address **259 Kent Avenue
Ferndale, 2194**
Telephone **+27 (0)11 781 5689**

Laboratory Manager **Mark Baird (acting)**
SGS Reference **JB11-01870 R0**
Report Number **0000001540**
Date Received **2011/09/12 11:20:06AM**
Date Reported **2011/10/03 11:26:35AM**

COMMENTS

The document is issued in accordance with SANAS's accreditation requirements.
Accredited for compliance with ISO/IEC 17025. SANAS accredited laboratory T0107.



This report/certificate is a re-issued copy and replaces the originally issued document dated 2011-09-30. The reason for re-issue is that percent solids results were omitted from the original report.

Filter cake samples not dried prior to testing.

Sample(s) leached using deionised water. Results reported on leachate.

SIGNATORIES

Gladness Radebe
Technical Supervisor/Technical Signatory

Sarah Newton
Technical Consultant/Technical Signatory



ANALYTICAL REPORT

JB11-01870 R0

Report number 0000001540
 Client reference: 11521198

Sample Number	JB11-01870.001	JB11-01870.002
Sample Name	Ashing Ash	Dusting Ash
Sample Matrix	Ash sample	Ash sample

Parameter	Units	LOR		
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Moisture Method:

Parameter	Units	LOR		
Solids content*	%	0.050	6.37	48.3

South African Standard Leach Procedure Method: AS 4439.3

Final pH	-	-	10.9	11.8
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Conductivity - Water Method: ME-ANA-AN-007

Conductivity	mS/m	2.0	24	160
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Total Dissolved Solids (TDS) in water Method: ME-ANA-AN-011

Total Dissolved Solids	mg/l	21.0	64	272
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Anions by Ion Chromatography Method: ME-ANA-AN-AN014

Fluoride	mg/l	0.050	<0.050	0.30
Chloride	mg/l	0.050	1.7	2.1
Nitrate	mg/l	0.10	0.28	1.5
Sulphate	mg/l	0.050	19	13

Hexavalent Chromium by UV-VIS Method: ME-ANA-AN-018

Hexavalent Chromium*	mg/l	0.010	<0.010	0.11
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Ammonia as N by UV Method: APHA4500_NH3

Ammonia*	mg/l	0.050	<0.050	<0.050
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ICP-OES Metals in Water (Dissolved) Method: ME-ANA-AN-027

Silver	mg/l	0.0020	<0.0020	<0.0020
Aluminium	mg/l	0.020	1.6	4.4
Boron	mg/l	0.0050	0.39	0.20
Barium	mg/l	0.0020	0.059	0.84
Beryllium	mg/l	0.00010	<0.00010	<0.00010
Calcium	mg/l	0.50	28	130
Iron	mg/l	0.050	<0.050	<0.050
Potassium	mg/l	0.20	0.45	1.0
Lithium	mg/l	0.0050	0.011	0.068
Magnesium	mg/l	0.010	0.46	0.018
Sodium	mg/l	0.50	3.5	5.0
Silicon	mg/l	1.0	7.1	4.3
Strontium	mg/l	0.0010	0.41	2.1
Titanium	mg/l	0.0050	<0.0050	<0.0050
Vanadium	mg/l	0.0010	0.022	0.045
Zinc	mg/l	0.010	<0.010	<0.010

ICP-MS Metals (Dissolved) Method: ME-ANA-AN-026

Arsenic	mg/l	0.0030	0.012	<0.0030
Bismuth	mg/l	0.0010	0.0020	<0.0010
Cadmium	mg/l	0.0020	0.0024	<0.0020
Cobalt	mg/l	0.0020	0.0027	<0.0020
Chromium	mg/l	0.0030	0.0075	0.11
Copper	mg/l	0.0040	<0.0040	<0.0040
Mercury	mg/l	0.00010	0.00015	0.00030
Manganese	mg/l	0.0030	0.0097	<0.0030
Molybdenum	mg/l	0.0070	0.012	0.067
Nickel	mg/l	0.0070	<0.0070	<0.0070



ANALYTICAL REPORT

JB11-01870 R0

Report number: 0000001540
Client reference: 11521198

Sample Number	JB11-01870.001	JB11-01870.002
Sample Name	Ashing Ash	Dusting Ash
Sample Matrix	Ash sample	Ash sample

Parameter	Units	LOR		
ICP-MS Metals (Dissolved) Method: ME-ANA-AN-026 (continued)				
Lead	mg/l	0.0040	<0.0040	<0.0040
Antimony	mg/l	0.0070	<0.0070	<0.0070
Selenium	mg/l	0.0040	<0.0040	<0.0040
Tin	mg/l	0.0070	<0.0070	<0.0070

METHOD

METHODOLOGY SUMMARY

FOOTNOTES

IS Insufficient sample for analysis.
LNR Sample listed, but not received.
• This analysis is not covered by the scope of accreditation.
^ Performed by outside laboratory.
LOR Limit of Reporting
↑↓ Raised or Lowered Limit of Reporting

QFH QC result is above the upper tolerance
QFL QC result is below the lower tolerance
- The sample was not analysed for this analyte

Samples analysed as received.
Solid samples expressed on a dry weight basis.

Unless otherwise indicated, samples were received in containers fit for purpose.

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ZITHOLELE CONSULTING (PTY) LTD

WASTE CLASSIFICATION OF POWER STATION
ASH AND BRINE FROM THE CAMDEN POWER STATION

Report: JW164/11/D116 - REV

Appendix B

CHEMICAL ANALYSES CONDUCTED ON THE REVERSE OSMOSIS BRINE



Central Water Laboratory

Final Task Report

Report Reference

WL2012-010199

Attention	Irma Hodgskin
Client Name	Camden Power Station
Address	Piet Retief Road
Fax	017 827 8115
Telephone	017 827 8007

Date 2012/06/25
Tel. No. +27 11 629 5596
Fax. No. +27 11 629 5528

Report Title WMC

TEST RESULTS FOR THE ANALYSIS OF WATER
SAMPLES

Number of Samples 5

Description of Samples ACCEPTABLE

Date Registered 12-June-2012

Date Reported 25-June-2012

Task Comments:

Approved By : _____

Cody Makhuba

Snr Technician

011 629 5596

Date : _____

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Eskom Central Water Laboratory Test Results



Laboratory Number T0055

Sample ID	3860564	WMC-2012-06-12/91	WL2012-010199
Brine water sample			
Component	Unit		Value
Alkalinity Total	mg/l CaCO ₃		26.7
Aluminium as Al	mg/l		0.67
Barium as Ba	mg/l		0.30
Calcium as Ca	mg/l		480
Chloride as Cl	mg/l		330.00
Iron as Fe	mg/l		<0.005
Flouride as F	mg/l		3.64
Magnesium as Mg	mg/l		0.73
Manganese as Mn	mg/l		0.01
Sodium as Na	mg/l		420
Nitrate as N	mg/l		3.14
pH @ 25 °C			6.54
Ortho Phosphate as PO ₄	mg/l		<0.090
Silica as SiO ₂	mg/l		18
Sulphate as SO ₄	mg/l		2100
Strontium as Sr	mg/l		13

Sample ID	3860565	WMC-2012-06-12/92	WL2012-010199
Raw Feed water sample			
Component	Unit		Value
Alkalinity Total	mg/l CaCO ₃		132
Aluminium as Al	mg/l		0.94
Barium as Ba	mg/l		0.18
Calcium as Ca	mg/l		180
Chloride as Cl	mg/l		130.00
Iron as Fe	mg/l		<0.005
Flouride as F	mg/l		1.47
Magnesium as Mg	mg/l		0.18
Manganese as Mn	mg/l		0.01
Sodium as Na	mg/l		170
Nitrate as N	mg/l		1.35
pH @ 25 °C			10.52
Ortho Phosphate as PO ₄	mg/l		<0.090
Silica as SiO ₂	mg/l		8.8
Sulphate as SO ₄	mg/l		610
Strontium as Sr	mg/l		4.8

Eskom Central Water Laboratory Test Results



Laboratory Number T0055

Sample ID	3860566	WMC-2012-06-12/93	WL2012-010199
Maddox Out water sample			
Component	Unit	Value	
Alkalinity Total	mg/l CaCO ₃	13.6	
Aluminium as Al	mg/l	0.21	
Barium as Ba	mg/l	0.10	
Calcium as Ca	mg/l	180	
Chloride as Cl	mg/l	130.00	
Iron as Fe	mg/l	0.02	
Flouride as F	mg/l	1.48	
Magnesium as Mg	mg/l	0.24	
Manganese as Mn	mg/l	0.01	
Sodium as Na	mg/l	160	
Nitrate as N	mg/l	1.36	
pH @ 25 °C		5.97	
Ortho Phosphate as PO ₄	mg/l	<0.090	
Silica as SiO ₂	mg/l	7.2	
Sulphate as SO ₄	mg/l	730	
Strontium as Sr	mg/l	4.8	

Sample ID	3860567	WMC-2012-06-12/94	WL2012-010199
Gac Out water sample			
Component	Unit	Value	
Alkalinity Total	mg/l CaCO ₃	15.1	
Aluminium as Al	mg/l	0.25	
Barium as Ba	mg/l	0.14	
Calcium as Ca	mg/l	180	
Chloride as Cl	mg/l	130.00	
Iron as Fe	mg/l	<0.005	
Flouride as F	mg/l	1.46	
Magnesium as Mg	mg/l	0.25	
Manganese as Mn	mg/l	0.01	
Sodium as Na	mg/l	160	
Nitrate as N	mg/l	1.35	
pH @ 25 °C		5.81	
Ortho Phosphate as PO ₄	mg/l	<0.090	
Silica as SiO ₂	mg/l	7.4	
Sulphate as SO ₄	mg/l	730	
Strontium as Sr	mg/l	4.7	

Eskom Central Water Laboratory Test Results



Laboratory Number T0055

Sample ID	3860568	WMC-2012-06-12/95	WL2012-010199
Pertmate Product			
water sample			
Component	Unit	Value	
Alkalinity Total	mg/l CaCO ₃	3.7	
Aluminium as Al	mg/l	0.04	
Barium as Ba	mg/l	<0.005	
Calcium as Ca	mg/l	1.9	
Chloride as Cl	mg/l	5.28	
Iron as Fe	mg/l	<0.005	
Flouride as F	mg/l	0.08	
Magnesium as Mg	mg/l	<0.005	
Manganese as Mn	mg/l	0.01	
Sodium as Na	mg/l	6.9	
Nitrate as N	mg/l	0.27	
pH @ 25 °C		5.75	
Ortho Phosphate as PO ₄	mg/l	<0.090	
Silica as SiO ₂	mg/l	0.16	
Sulphate as SO ₄	mg/l	5.59	
Strontium as Sr	mg/l	0.05	

The analyses were performed using the following methods:

Alkalinity Total	ESKOM METHOD NO 304	Accredited
Aluminium ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Barium ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Calcium ICP (mg/l)	ESKOM METHOD NO 415	Accredited
Chloride IC (mg/l)	ESKOM METHOD NO 307	Accredited
Flouride IC (mg/l)	ESKOM METHOD NO 307	Not Accredited
Iron ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Magnesium ICP (mg/l)	ESKOM METHOD NO 415	Accredited
Manganese ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Nitrate as N IC (mg/l)	ESKOM METHOD NO 307	Accredited
Ortho Phosphate as PO₄(mg/l)	ESKOM METHOD NO 72	Not Accredited
pH @ 25 °C	ESKOM METHOD NO 300A	Accredited
Silica as SiO₂ ICP (mg/l)	ESKOM METHOD NO 417	Not Accredited
Sodium ICP (mg/l)	ESKOM METHOD NO 415	Accredited
Strontium ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Sulphate IC (mg/l)	ESKOM METHOD NO 307	Accredited

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4. With the exception of all microbiological analyses, unless otherwise stated, sampling is not carried out by the laboratory.
5. All water samples are preserved according to procedure P511 unless otherwise stated.
6. Unless otherwise specified all analyses on water samples give the dissolved constituents.

End of the Report

Central Water Laboratory

Final Task Report

Report Reference

WL2012-010221

Attention	Irma Hodgskin
Client Name	Camden Power Station
Address	Piet Retief Road
Fax	017 827 8115
Telephone	017 827 8007

Date 2012/07/09
Tel. No. +27 11 629 5596
Fax. No. +27 11 629 5528

Report Title WMC

TEST RESULTS FOR THE ANALYSIS OF WATER
SAMPLES

Number of Samples 1

Description of Samples Acceptable

Date Registered 29-June-2012

Date Reported 09-July-2012

Task Comments:

Approved By : _____

Cody Makhuba

Snr Technician

011 629 5596

Date : _____

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Opinions and interpretations expressed herein are outside the scope of SANAS accreditation,

PLEASE NOTE: The test results relate only to the specified samples tested as identified in this report.

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Eskom Central Water Laboratory Test Results



Laboratory Number T0055

Sample ID	3906049	WMC-2012-06-29/13	WL2012-010221
SAMPLE 1			
RO BRINE			
Component	Unit	Value	
Alkalinity Total	mg/l CaCO ₃	29.1	
Aluminium as Al	mg/l	0.07	
Barium as Ba	mg/l	0.25	
Calcium as Ca	mg/l	640	
Chloride as Cl	mg/l	380.00	
Iron as Fe	mg/l	0.01	
Flouride as F	mg/l	3.47	
Magnesium as Mg	mg/l	0.60	
Manganese as Mn	mg/l	<0.005	
Sodium as Na	mg/l	570	
Nitrate as N	mg/l	3.32	
pH @ 25 °C		7.12	
Ortho Phosphate as PO ₄	mg/l	<0.090	
Silica as SiO ₂	mg/l	22	
Sulphate as SO ₄	mg/l	2080	
Strontium as Sr	mg/l	15	

The analyses were performed using the following methods:

Alkalinity Total	ESKOM METHOD NO 304	Accredited
Aluminium ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Barium ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Calcium ICP (mg/l)	ESKOM METHOD NO 415	Accredited
Chloride IC (mg/l)	ESKOM METHOD NO 307	Accredited
Flouride IC (mg/l)	ESKOM METHOD NO 307	Not Accredited
Iron ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Magnesium ICP (mg/l)	ESKOM METHOD NO 415	Accredited
Manganese ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Nitrate as N IC (mg/l)	ESKOM METHOD NO 307	Accredited
Ortho Phosphate as PO ₄ (mg/l)	ESKOM METHOD NO 72	Not Accredited
pH @ 25 °C	ESKOM METHOD NO 300A	Accredited
Silica as SiO ₂ ICP (mg/l)	ESKOM METHOD NO 417	Not Accredited
Sodium ICP (mg/l)	ESKOM METHOD NO 415	Accredited
Strontium ICP (mg/l)	ESKOM METHOD NO 412	Accredited
Sulphate IC (mg/l)	ESKOM METHOD NO 307	Accredited

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4. With the exception of all microbiological analyses, unless otherwise stated, sampling is not carried out by the laboratory.
5. All water samples are preserved according to procedure P511 unless otherwise stated.
6. Unless otherwise specified all analyses on water samples give the dissolved constituents.

End of the Report